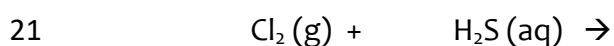
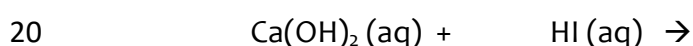
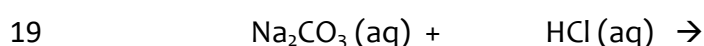
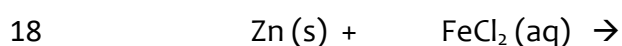
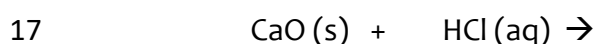
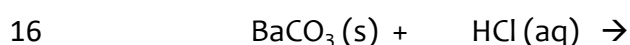
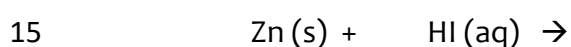
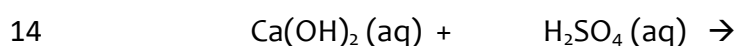
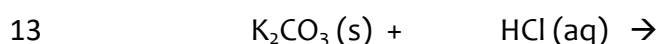
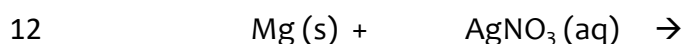
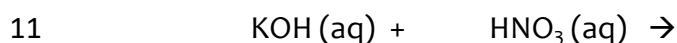
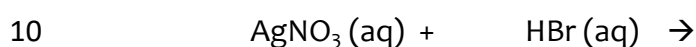
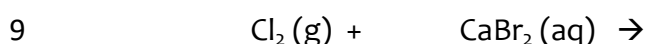
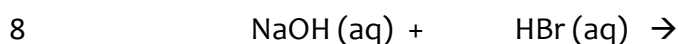
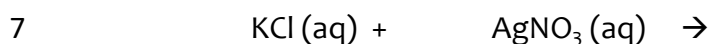
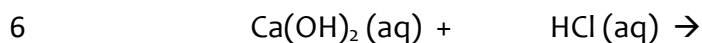
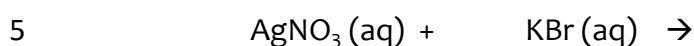
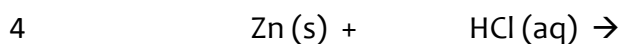
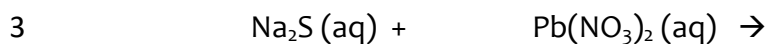
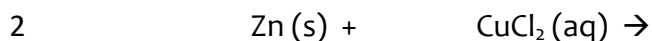
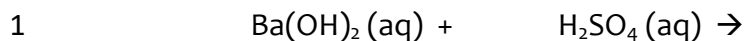
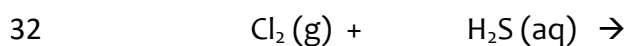
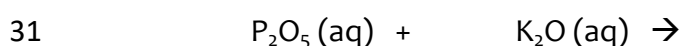
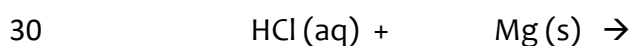
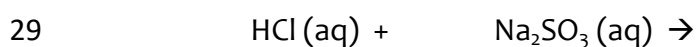
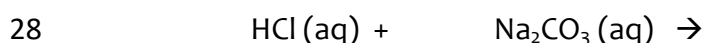
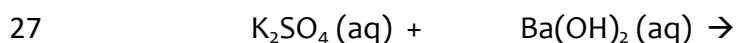
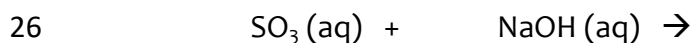
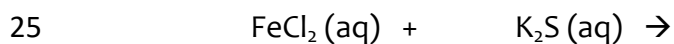
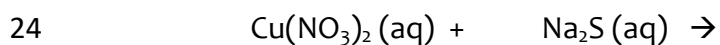
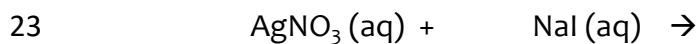
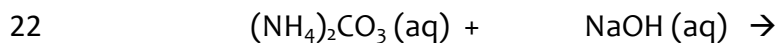


ΧΗΜΕΙΑ Α' ΛΥΚΕΙΟΥ - Φύλλο Εργασίας: Χημικές Αντιδράσεις

ΗΜΕΡΟΜ: ΟΝΟΜΑΤΕΠΩΝΥΜΟ: ΤΜΗΜΑ:

Να συμπληρώσετε τις αντιδράσεις με τα προϊόντα και τους σωστούς συντελεστές





- 33 i. ii. Να συμπληρώσετε τα προϊόντα και τους συντελεστές της αντίδρασης:
- $(\text{NH}_4)_2\text{S} (\text{aq}) + \text{NaOH} (\text{aq}) \rightarrow$ _____
- iii. Ονομάστε τις χημικές ενώσεις: NaOH , $(\text{NH}_4)_2\text{S}$, Na_2S & NH_3
- iv. Υπολογίστε τον Αριθμό Οξείδωσης του C στα εξής: CO_3^{-2} , CO & CO_2
-

ΣΗΜΕΙΩΣΗ: Σε όλες τις παραπάνω αντιδράσεις να αιτιολογήσετε τον λόγο που πραγματοποιούνται.
