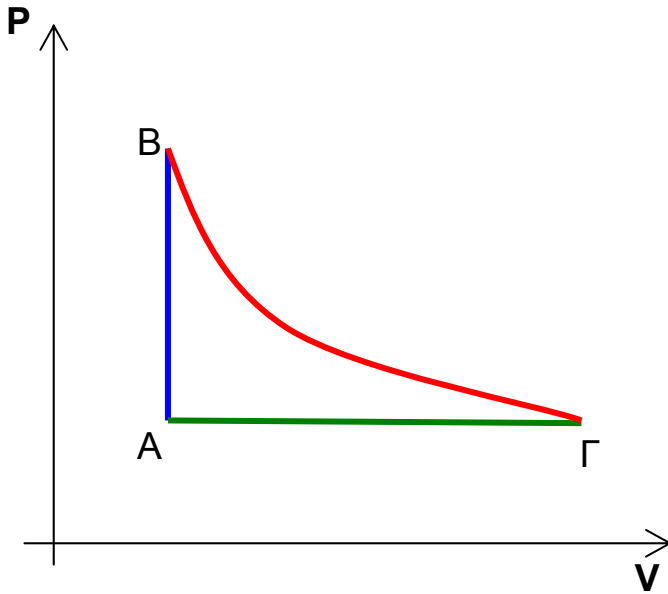


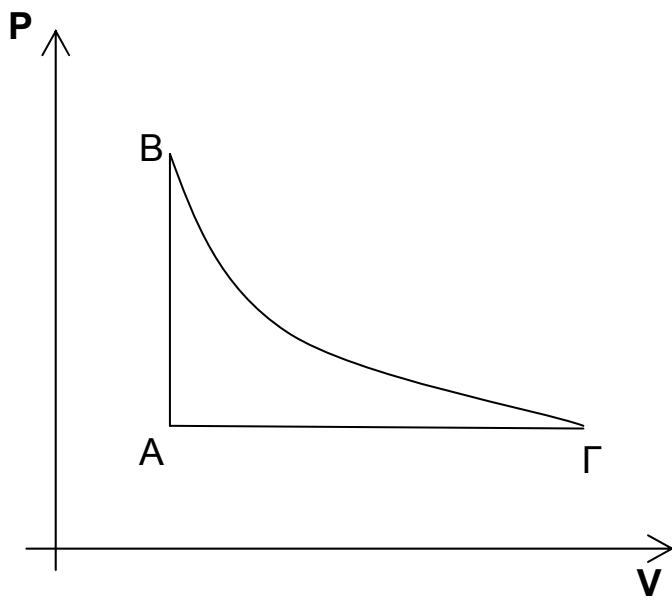
Μεταβολές των Αερίων

είδη μεταβολών:



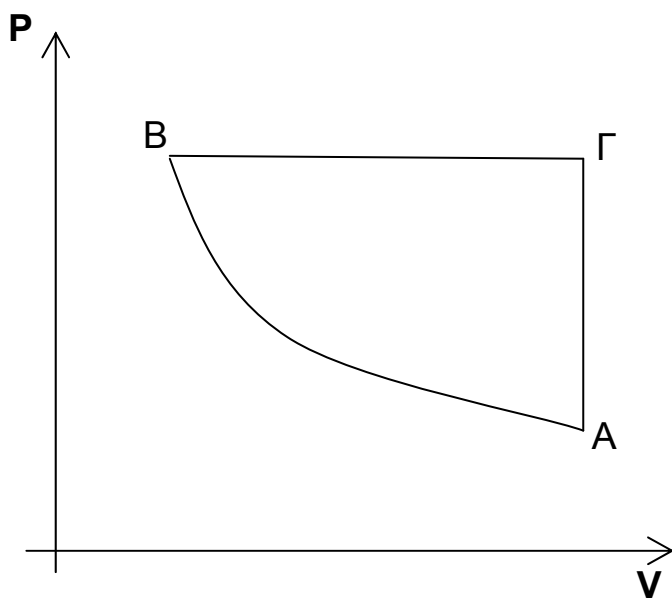
Ονομασία	Σταθερό	Μονάδες	Τύπος
ισόχωρη	Όγκος (V)	m ³	$\frac{P_1}{P_2} = \frac{T_1}{T_2}$
ισόθερμη	Θερμοκρασία (T)	°C (σε K: +273)	$\frac{P_1}{P_2} = \frac{V_2}{V_1}$
ισοβαρής	Πίεση (P)	atm - bar	$\frac{V_1}{V_2} = \frac{T_1}{T_2}$

Άσκηση 1



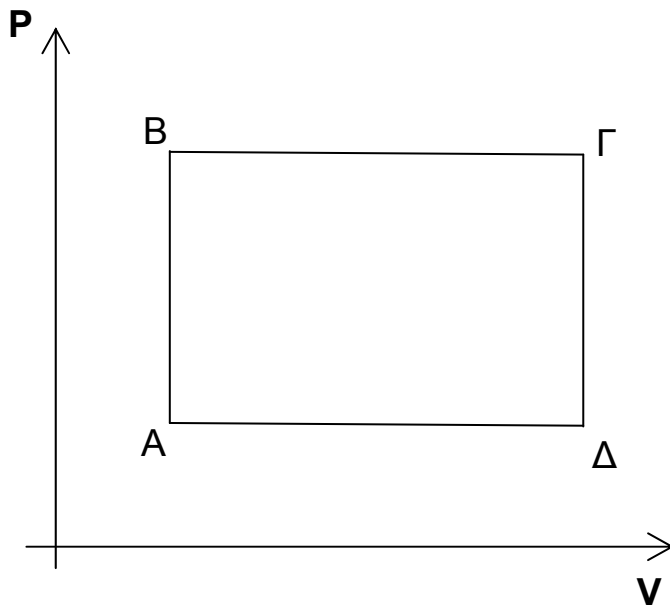
	P	V	T
A			27
B	2	1	
Γ	1		

Άσκηση 2



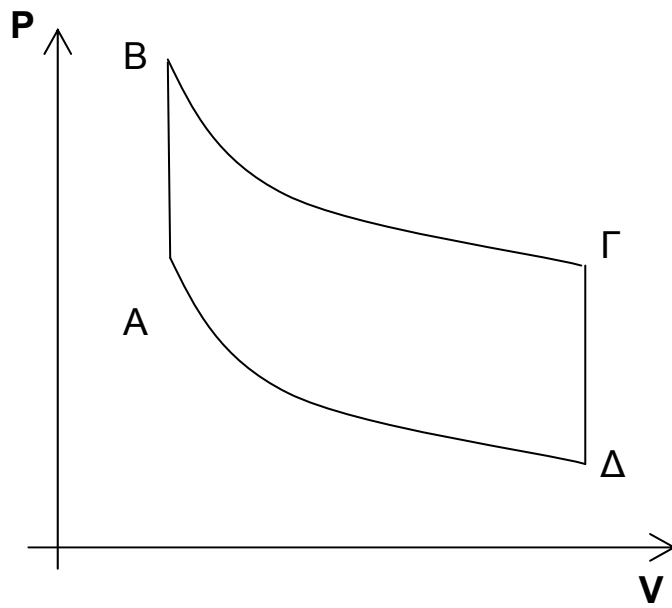
	P	V	T
A	1		27
B		1	
Γ	2		

Άσκηση 3



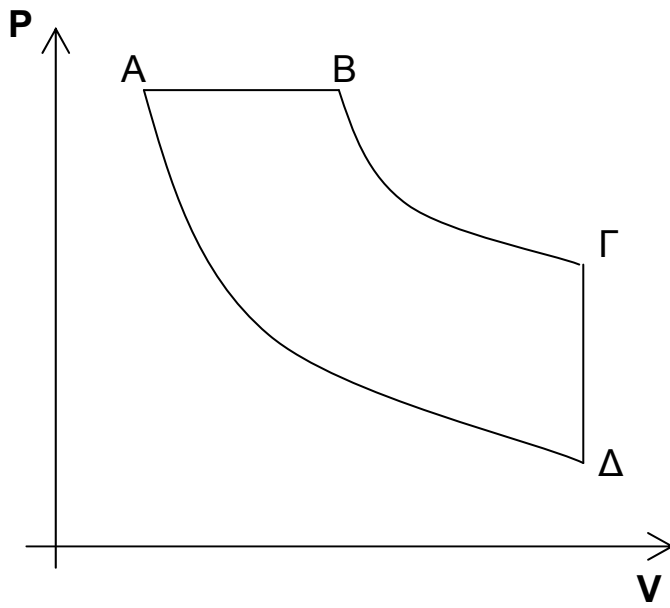
	P	V	T
A	1		27
B	2	1	
Γ		2	
Δ			

Άσκηση 4



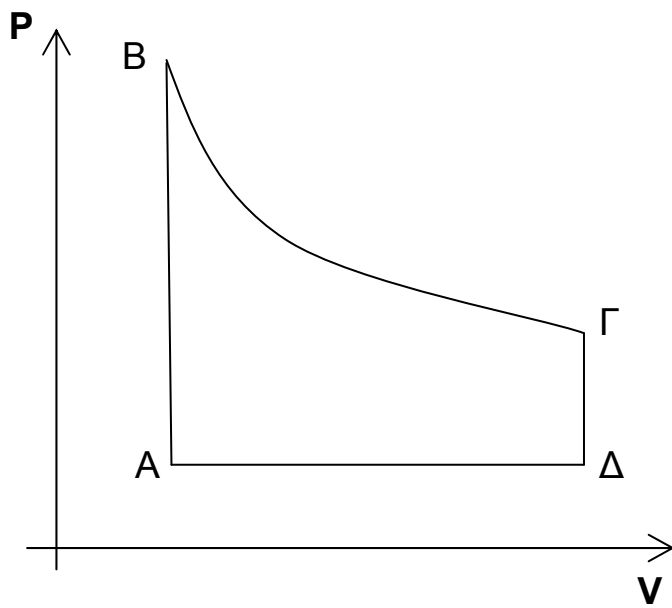
	P	V	T
A	1		27
B	2	1	
Γ			
Δ		2	

Άσκηση 5



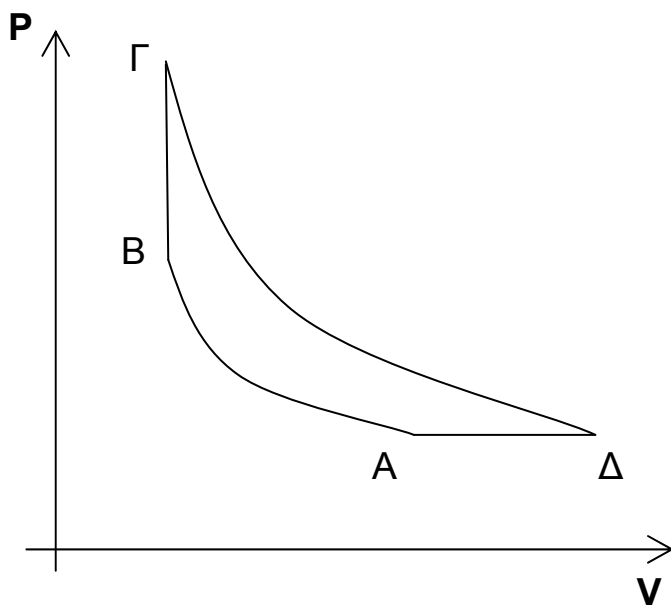
	P	V	T
A	4	1	
B		2	127
Γ	2		
Δ			

Άσκηση 6



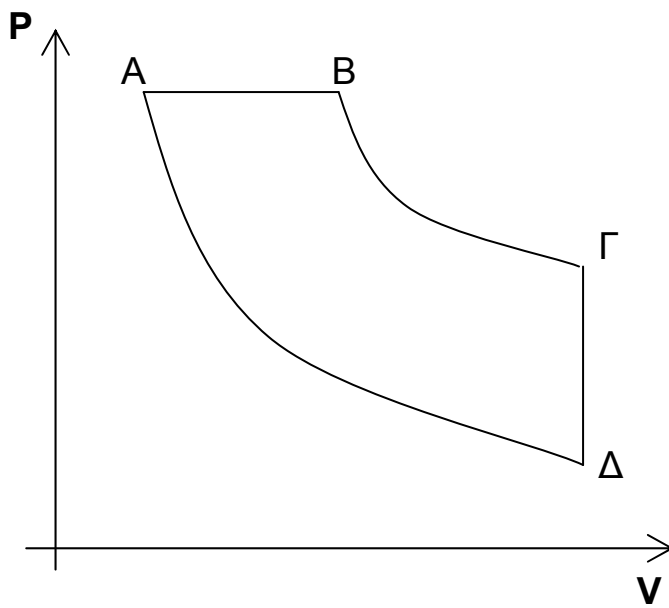
	P	V	T
A	1		
B	2		
Γ			127
Δ		3	27

Άσκηση 7



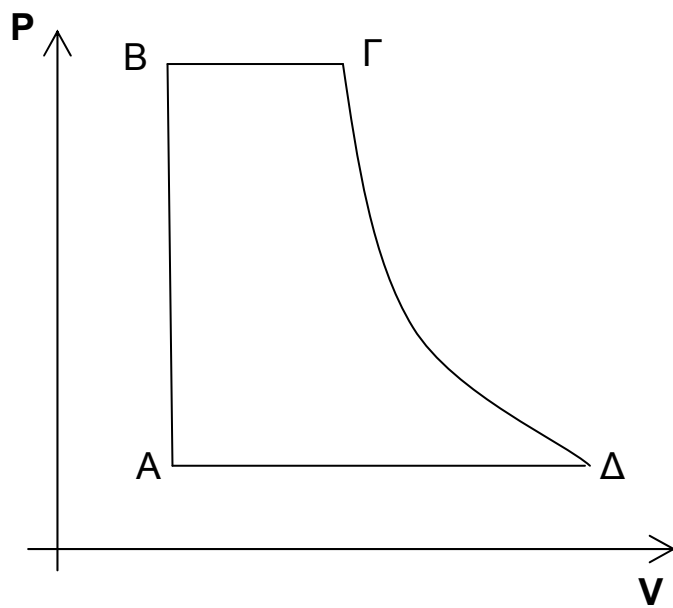
	P	V	T
A	1		27
B		1	
Γ			127
Δ		3	

Άσκηση 8



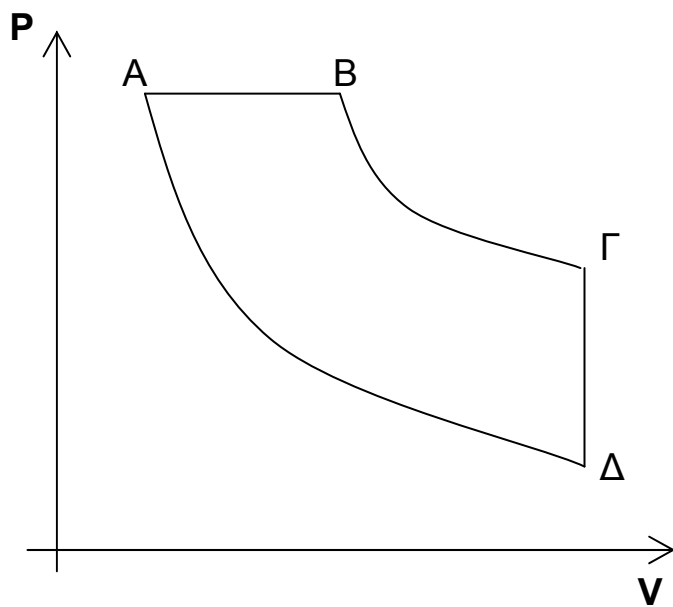
	P	V	T
A	4		
B			227
Γ	2		
Δ		4	27

Άσκηση 9



	P	V	T
A	1	1	
B	3		327
Γ		2	
Δ			

Άσκηση 10



	P	V	T
A	1	10	23
B	10		
Γ		2	
Δ			